Meitnerium



General Information

Discovery

Meitnurium was first made in 1982 by Peter Armbruster, Gottfried Munzenberg and co-workers at the GSI in Darmstadt, Germany.

Appearance

Unknown, but probably metallic grey in appearance.

Source

A transuranium element, less than 10 atoms of meitnerium have ever been made, and it will probably never be isolated in observable quantities. Created by a so-called "cold fusion" method, in which a target of bismuth is bombarded with atoms of iron.

Uses

Unknown.

Biological Role

None.

General Information

A synthetic element created via nuclear bombardment, few atoms have ever been made and the properties of meitnerium are very poorly understood. It is a radioactive metal which does not occur naturally and is of research interest only. The first atoms were made via a nuclear reaction, the cold fusion method:

 209 Bi + 58 Fe ightarrow 266 Mt + n

Physical Information

Atomic Number 109

Relative Atomic Mass (¹²C=12.000) 266

Melting Point/K Not available

Boiling Point/K Not available

Density/kg m⁻³ Not available

Ground State Electron Configuration [Rn]5f¹⁴6d⁷7s²

Electron Affinity (M-M⁻)/kJ mol⁻¹ Not available

Key Isotopes

Nuclide ²⁶⁶Mt

Atomic mass 266.14

Natural abundance 0%

Half-life approx

3.4x10⁻³secs

Ionisation Energies/kJ mol -1

M - M⁺ 840 (est)

M⁺ - M²⁺

 $M^{2+} - M^{3+}$

 M^{3+} - M^{4+}

 $M^{4+} - M^{5+}$

M⁵⁺ - M⁶⁺

IVI - IVI

 $M^{6+} - M^{7+}$

 $M^{7+} - M^{8+}$

 $M^{8+} - M^{9+}$

 $M^{9+} - M^{10+}$

Other Information

Enthalpy of Fusion/kJ mol⁻¹ Not available

Enthalpy of Vaporisation/kJ mol⁻¹ Not available

Oxidation States

Mt^{II} has been predicted as probably the most stable state.

Covalent Bonds/kJ mol⁻¹

Not available