# **Einsteinium**



## **General Information**

#### Discovery

Einsteinium was discovered by G.R. Choppin, S.G. Thompson, A Ghiorso and B.G. Harvey in 1952, in the debris of the thermonuclear explosion in the Pacific at Eniwetok. This involved the examination of tons of radioactive coral from the blast area.

### Appearance

Einsteinium is a radioactive, silvery metal.

#### Source

Einsteinium can be obtained in milligram quantities from the neutron bombardment of plutonium.

#### Uses

Einsteinium has no uses outside research.

### **Biological Role**

Einsteinium has no known biological role. It is toxic due to its radioactivity.

### **General Information**

Einsteinium is attacked by oxygen, steam and acids but not by alkalis.

## **Physical Information**

Atomic Number	99
Relative Atomic Mass ( <sup>12</sup> C=12.000)	254 (radioactive)
Melting Point/K	Not available
Boiling Point/K	Not available
Density/kg m <sup>-3</sup>	Not available
Ground State Electron Configuration	[Rn]5f <sup>11</sup> 7s <sup>2</sup>
Electron Affinity (M-M <sup>-</sup> )/kJ mol <sup>-1</sup>	50

# Key Isotopes

Nuclide	<sup>253</sup> ES	<sup>254</sup> Es	
Atomic mass		254.09	
Natural abundance	0%	0%	
Half-life	20.7 days	201 days	

## Ionisation Energies/kJ mol <sup>-1</sup>

М	- M <sup>+</sup>	619
$M^+$	- M <sup>2+</sup>	
M <sup>2+</sup>	- M <sup>3+</sup>	
M <sup>3+</sup>	- M <sup>4+</sup>	
$M^{4+}$	- M <sup>5+</sup>	
M <sup>5+</sup>	- M <sup>6+</sup>	
M <sup>6+</sup>	- M <sup>7+</sup>	
M <sup>7+</sup>	- M <sup>8+</sup>	
M <sup>8+</sup>	- M <sup>9+</sup>	
M <sup>9+</sup>	- M <sup>10+</sup>	

## **Other Information**

Enthalpy of Fusion/kJ mol <sup>-1</sup>	Not available	
Enthalpy of Vaporisation/kJ mol <sup>-1</sup>	Not available	
Oxidation States		
Main	Es <sup>™</sup>	
Others	Es"	
Covalent Bonds/kJ mol <sup>-1</sup>		
Not applicable		